# MO 13: Ultrafast Dynamics II

Time: Wednesday 11:00–13:00 Location: HS XVI

MO 13.1 Wed 11:00 HS XVI

Photo-induced structural dynamics of thymine with ultra- ${\rm fast\ electron\ diffusion} - {\bullet} {\rm X}$ iaojun Wang $^1,$  Jackson Lederer $^2,$ Dennis Mayer<sup>1</sup>, Fabiano Lever<sup>1</sup>, Pedro Nunes<sup>2</sup>, Yusong Liu<sup>3</sup>, Surjendu Bhattacharyya<sup>3</sup>, Nikhil Pachisia<sup>2</sup>, Xinxin Cheng<sup>3</sup>, TIANZHE  $XU^3$ , Stephen Weathersby<sup>3</sup>, Randy Lemons<sup>3</sup>, Patrick Kramer<sup>3</sup>, Philipp Lenzen<sup>3</sup>, Ming-Fu Lin<sup>3</sup>, Kirk Larsen<sup>3</sup>, Fuhao  $J_I^3$ , Robert England<sup>3</sup>, Christina Hampton<sup>3</sup>, Brian Kaufman<sup>3</sup>, ALICE GREEN<sup>3</sup>, ALEXANDER REID<sup>3</sup>, TODD MARTINEZ<sup>3,4</sup>, THOMAS  $\mathrm{Wo}\textsubscript{LF}^3, \mathrm{M}\textsubscript{ARTIN}\ \mathrm{C}\textsubscript{ENTURION}^2, \mathrm{and} \ \mathrm{M}\textsubscript{ARKUS}\ \mathrm{G}\textsubscript{U}\textsubscript{E}\textsubscript{H}^{1,5}$  —  $^1\mathrm{Deutsches}$ Elektronen-Synchrotron DESY, Hamburg, Germany — <sup>2</sup>Department of Physics and Astronomy, University of Nebraska-Lincoln, Lincoln, USA — <sup>3</sup>SLAC National Accelerator Laboratory, Menlo Park, USA — <sup>4</sup>Department of Chemistry, Stanford University, Stanford, USA — 5 Institute of Physical Chemistry, University of Hamburg, Hamburg, Germany

The photoprotection mechanism of thymine involves complex relaxation dynamics, where energy from photoexcitation is converted into vibrational energy through radiationless transitions. Previous studies have indicated that isolated thymine molecules undergo  ${}^1\pi\pi^*\rightarrow{}^1n\pi^*$ internal conversion within 100 femtoseconds (fs), followed by the intersystem crossing from  $1n\pi^*$  to the triplet state within 10 picoseconds (ps). However, the nuclear dynamics accompanying the transitions have not been experimentally resolved. Here we present a direct observation of the ultrafast nuclear motions using ultrafast electron diffraction, on the fs timescale and sub-Ångstrom resolution.

#### MO 13.2 Wed 11:15 HS XVI

Ultrafast electron diffraction imaging of wavelengthdependent trans-to-cis isomerization in azobenzene — Sur-JENDU BHATTARCHARYYA<sup>1</sup>, MING-FU LIN<sup>1</sup>, THOMAS J. A. WOLF<sup>1</sup>, ALICE E. GREEN<sup>1</sup>, YUSONG LIU<sup>1</sup>, XINXIN CHENG<sup>1</sup>, PHILIPP Lenzen<sup>1</sup>, Xiaozhe Shen<sup>1</sup>, •Kasra Amini<sup>2</sup>, Fernando Rodriguez<br>Diaz<sup>2</sup>, Fabiano Lever<sup>3</sup>, Xiaojun Wang<sup>3</sup>, Markus Guehr<sup>3</sup>, Mike  $\text{Minirtr1}^1, \text{Joel } \text{England}^1, \text{and } \text{Alexander H. }\text{RED}^1 \text{---}^1 \text{SLAC Na-}$ tional Accelerator Laboratory, Menlo Park, CA, USA  $-2$  Max-Born-Institut, Max-Born-Straße 2A, 12489, Berlin, Germany — <sup>3</sup>Deutsches Elektronen-Synchrotron DESY, Hamburg, Germany

Molecular photoswitches, exemplified by azobenzene (AB), represent a key class of molecules displaying optical switching and structural rearrangement around a double bond - a process that is fundamental to human vision. Ab initio calculations predict that optical excitation of trans-AB to its second excited state at 297-nm undergoes trans-to-cis isomerization with a quantum yield of 12% within a 350-fs timescale. However, experimental investigations into the photochemistry and nuclear dynamics of gas-phase AB remain limited. Here, we present an ultrafast electron diffraction (UED) study of trans-to-cis isomerization in AB between 267-nm and 330-nm using the MeV-UED instrument at SLAC. We show that the onset of isomerization is wavelengthdependent, and we measure the first direct observation of the transient cis isomer which is formed within the predicted 350-fs timescale.

## MO 13.3 Wed 11:30 HS XVI

Semiclassical simulations of laser-induced electron diffraction  $-$  • Álvaro Fernández<sup>1,2</sup>, Andrey Yachmenev<sup>1,3</sup>, and Jochen  $K$ üpper $1,2,3$  — <sup>1</sup>Deutsches Elektronen-Synchrotron, DESY, Hamburg  $^{-2}$ Department of Physics, Universität Hamburg —  $^{3}$ Center for Ultrafast Imaging, Universität Hamburg

Laser-induced electron diffraction (LIED) [1] is a tabletop imaging technique capable of measuring photoelectron probability densities, which allow for determining nuclear positions with sub-bondlength and femtosecond precision. In LIED, electrons have much lower energies than those of standard electron diffraction and are significantly perturbed by interactions with the parent ion during rescattering. Consequently, retrieving molecular structures from LIED requires accurate theoretical simulations of the complex photoelectron dynamics.

We present the implementation and results of semiclassical simulations of LIED [2]. In our model, the ionisation step is computed quantum-mechanically using the MO-ADK or Dyson formalisms. Subsequent photoelectron propagation and rescattering are treated classically, modelling the electrostatic potential of the parent ion using chosen quantum-chemical method. To obtain accurate results, billions

of different electron trajectories are simulated. The approach is implemented as a Python package, and its computational performance and accuracy are validated in simulations of OCS and indole-water molecules.

[1] Karamatskos, E. T, et al., J. Chem. Phys., 150, 24 (2019)

[2] Wiese, J., et al., Phys. Rev. Research, 3, 013089, (2021)

MO 13.4 Wed 11:45 HS XVI Towards Ultrafast Molecular Dynamics in Chiral Molecules in a Micro-solvated Environment — ∙Liliana M. Ramos Moreno<sup>1</sup>, Deepak K. Pandey<sup>1</sup>, Sagnik Das<sup>1</sup>, Claus-Peter<br>Schulz<sup>2</sup>, Hendrike Braun-Knie<sup>1</sup>, and Jochen Mikosch<sup>1</sup> — 1 Institut für Physik, Universität Kassel, Kassel, Germany — <sup>2</sup>Max Born Institute (MBI), Berlin, Germany

The environment surrounding a molecule, its solvent, affects the molecular properties and reaction dynamics. Solution in water is of particular relevance in biology and chemistry. Water is also of great fundamental interest as the "universal solvent", with the ability to dissolve more substances than most other liquids. We follow a bottom-up approach to investigate solution effects with femtosecond laser spectroscopy stepby-step, by adding water molecules one at a time to a small molecule in the gas phase. The clusters are produced in a temperature-variable molecular beam source[1]. The experiments are performed in a Photoelectron Photoion Coincidence (PEPICO) spectrometer. It consists of a magnetic bottle for photoelectrons and a Wiley-McLaren spectrometer for photoions, allowing us to retrieve time-of-flight spectra, in coincidence. We will present our experimental setup and the progress of an experiment where we study the time-dependent Circular Dichroism (CD) in 3-methyl cyclopentanone (3-MCP). The temperature-variable source allows us to change the conformer composition of 3-MCP and test a hypothesis established in recent work[2]. [1]Müller et al., JPCA 118, 8517 (2014) [2]Das et al., Control of circular dichroism in ion yield of 3-methyl cyclopentanone with femtosecond laser pulses(submitted).

#### MO 13.5 Wed 12:00 HS XVI

Dynamics of pyrrole-water studied by Coulomb-explosion  $\text{imaging}$  —  $\bullet$ SEBASTIAN TRIPPEL<sup>1,2</sup> and JOCHEN KÜPPER<sup>1,2,3</sup> — <sup>1</sup>Center for Free-Electron Laser Science CFEL, Deutches Elektronen-Synchrotron DESY, Hamburg  $-$  <sup>2</sup>Center for Ultrafast Imaging, Universität Hamburg — <sup>3</sup>Department of Physics, Universität Hamburg

Microsolvated biomolecules are promising model systems for studying the corresponding light-induced dynamics of molecules in solution [1]. Due to the still manageable complexity of the small clusters in the gas phase, atomic, molecular, and optical physics methods can be used to analyze and characterize their dynamics. Here, we will present our findings on the dynamics of ionized pyrrole-water [2]. Furthermore, we will discuss the creation of highly charged carbon atoms observed for the microsolvated clusters.

[1] L. He, et int. (8 authors), J. Küpper, J. Phys. Chem. Lett. 14, 10499 (2023)

[1] M. Johny, et int. (6 authors), J. Küpper, Phys. Chem. Chem. Phys. 26, 13118 (2024)

#### MO 13.6 Wed 12:15 HS XVI

Unraveling the dynamics of ionized water dimer in a highlypurified molecular beam —  $\bullet$ Ivo S. VINKLÁREK<sup>1</sup>, HUBERTUS<br>BROMBERGER<sup>1</sup>, LUISA BLUM<sup>1,2</sup>, SEBASTIAN TRIPPEL<sup>1</sup>, and JOCHEN  $K\ddot{\nu}$ PPER<sup>1,2,3</sup> — <sup>1</sup>Center for Free-Electron Laser Science CFEL, Deutsches Elektronen-Synchrotron DESY, Hamburg, Germany — <sup>2</sup>Department of Physics, Universität Hamburg, Hamburg, DE — <sup>3</sup>Center for Ultrafast Imaging, Universität Hamburg, Hamburg, DE

Radiation chemistry in biochemical systems is primarily driven by the ultrafast dynamics of water molecules after absorption of ionizing radiation. The initial water response to the ionization involves ultrafast hydrogen-bond-mediated proton transfer (PT), which was recently probed in the prototypical water-dimer cation  $(H_2O)_2$  [1], and subsequent fragmentation into highly reactant ions and radicals.

Our detailed study utilizing purified molecular beams of  $(H_2O)_2$  [2] revealed that  $(H_2O)_2^+$  can either stabilize or undergo fragmentation along more than ten distinct pathways. While theoretical studies have explored the rates and dynamics of some of these reactions, experimental evidence is completely lacking. To address this, we employed a disruptive-probing scheme [3] that allows us to track early PT dynamics and the populations of ionic products, thus directly yielding effective reaction-rate constants. These findings provide crucial insights into ionizing processes in both the atmosphere and living organisms.

[1] Schnorr, K. et al., Sci. Adv. 9, eadg7864 (2023)

[2] Vinklárek, I.S. et al., J. Phys. Chem. A 128, 1593 (2024),

[3] Jochim, B. et al., Rev. Sci. Instrum. 93, 033003 (2022)

### MO 13.7 Wed 12:30 HS XVI

Understanding fragmentation dynamics of difluorodi- $\mathbf{i}$ odomethane — •Nidin Vadassery<sup>1,3</sup>, Sebastian Trippel<sup>1,2,4</sup>, and JOCHEN  $K$ ÜPPER<sup>1,2,3,4</sup> — <sup>1</sup>Center for Free-Electron Laser Science, Deutches Elektronen-Synchrotron DESY, Hamburg, Germany  $2$ Center for Ultrafast Imaging, Universität Hamburg, Hamburg, Germany — <sup>3</sup>Department of Chemistry, Universität Hamburg, Hamburg, Germany — <sup>4</sup>Department of Physics, Universität Hamburg, Hamburg, Germany

Unimolecular photo-fragmentation is crucial in environmental chemical reactions like ozone depletion and aerosol formation [1]. Photodissociation of polyhalohydrocarbons significantly contributes to these climate-impacting processes. Difluorodiiodomethane  $(CF<sub>2</sub>I<sub>2</sub>)$  exhibits unconventional dynamics near dissociative energies [2]. We present experimental results exploring CF2I<sup>2</sup> dissociation dynamics using nearinfrared laser pulses. A pure  $CF<sub>2</sub>I<sub>2</sub>$  sample was produced using an electrostatic deflector, revealing metastable states and quantum-statespecific dynamics during photo-fragmentation [3]. We also outline a time-resolved x-ray diffraction approach and present computational results of diffraction simulations for laser-aligned gas-phase  $CF_2I_2$ .

[1] J. C. G. Martin, et al., J. Am. Chem. Soc. 144, 9240 (2022)

[2] P. Z. El-Khoury, et al., J. Chem. Phys. 132, 124501 (2010)

[3] I. S. Vinklárek, et int. (3 authors), J. Küpper, S. Trippel, J. Phys. Chem. A 128, 1593(2024) arXiv:2308.08006 [physics].

## MO 13.8 Wed 12:45 HS XVI Single and Double Ionization of Pyridine and Pyridine-water Complexes — •Sitanath Mondal<sup>1</sup>, Brendan Wouterlood<sup>1</sup>, Myriam Drissi<sup>2</sup>, Gustavo A. Garcia<sup>2</sup>, Laurent Nahon<sup>2</sup>, Frank STIENKEMEIER<sup>1</sup>, and SEBASTIAN HARTWEG<sup>1</sup> — <sup>1</sup>Institute of Physics, Albert-Ludwigs-Universität Freiburg, Germany — <sup>2</sup>Desirs Beamline, Synchrotron Soleil, St. Aubin, France

Cell and gene damage caused by slow electrons created by secondary processes after interaction with ionizing radiation is an important field of research. Photoelectron studies of biomolecule-water complexes in the gas phase can give us insight about some of the fundamental processes producing low energy secondary electrons and inducing fragmentation of cationic products. I will present a photoelectron photoion coincidence study of pyridine, pyridine clusters, and pyridine-water complexes considering single and double ionization processes. Our data provides insight on the single and double ionization energies of these systems and reveals differences in the dissociation pathways of pyridine in different environments. The ion-ion coincidence spectroscopy allows us to distinguish between different local double ionization and non-local double ionization mechanisms like intermolecular Coulombic Decay.